

A NOVEL SYNTHESIS OF HALF-PROTECTED 1,3-DICARBONYL COMPOUNDS

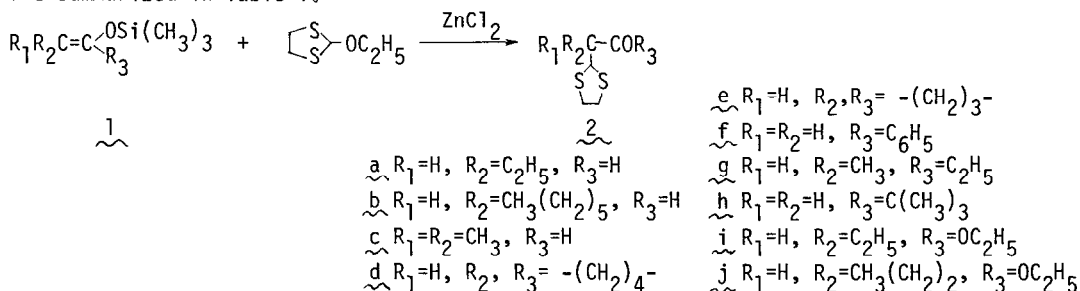
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*Summary:* It has been found that the reaction of enol silyl ethers with 2-ethoxy-1,3-dithiolane proceeds smoothly in the presence of zinc chloride catalyst to afford half-protected 1,3-dicarbonyl compounds in moderately good yields.

In 1959, it was shown that several  $\beta$ -chlorovinyl ketones are able to react with 1,2-ethanedithiol in the presence of potassium carbonate to give (1,3-dithiolan-2-yl)methyl ketones.<sup>1</sup> Recently, Taylor and his co-worker<sup>2</sup> described a process for the introduction of a base-stable protected formyl substituent to carbonyl compounds. This procedure involves the reaction of 2-chloro-1,3-dithiane with the morpholino enamines of the carbonyl compounds.

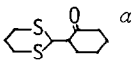
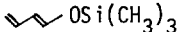
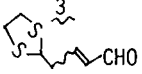
In the present communication, we show that enolizable aldehydes or ketones can be converted to the half-protected 1,3-dicarbonyl compounds 2 by reaction of their enol silyl ethers (1-alkenyl silyl ethers) 1 with 2-ethoxy-1,3-dithiolane in the presence of zinc chloride. The conversion can be effected most cleanly and efficiently using the zinc chloride catalyst, since the formation of 2,2'-[1,2-ethanediy]bis(thio)]bis-1,3-dithiolane<sup>3</sup> as a by-product from 2-ethoxy-1,3-dithiolane is avoided in contrast to the use of aluminum chloride, titanium(IV) chloride or boron trifluoride etherate. The yields of the products<sup>4</sup> under various conditions are summarized in Table I.



The following general experimental procedure is representative of this conversion. To a stirred, cooled (0°C) mixture of 2.4 mmol of the appropriate enol silyl ether and 0.30 g (2.0 mmol) of 2-ethoxy-1,3-dithiolane in 2 ml of acetonitrile or in 2 ml of dichloromethane (or without any solvent) is added during about 1-2 min 0.1 g (0.75 mmol) of zinc chloride. After addition is complete, the mixture is stirred at 0°C or room temperature (20-25°C) for such time till the complete consumption of the added 2-ethoxy-1,3-dithiolane which is ascertained by thin-layer chromatography. Then it is poured into 50 ml of cold brine, and extracted with several portions of ether. The combined extracts are dried (MgSO<sub>4</sub>), filtered, and

concentrated *in vacuo* to give a pale yellow colored residue. The product is isolated by column chromatography on silica gel with 5% ethyl acetate-hexane as the eluent. The results using 1 and 2-ethoxy-1,3-dithiolane were more satisfactory than Taylor's results<sup>2</sup> in view of the yield. Considering that both of the starting reagents are readily available, the present method provides one of the most convenient routes for the preparation of half-protected 1,3-dialdehydes and  $\beta$ -keto aldehydes.

Table 1. Reaction of Enol Silyl Ethers with 2-Ethoxy-1,3-dithiolane in the Presence of  $ZnCl_2$

| Enol silyl ether  | Product   | Reaction conditions(solvent/temperature,°C/time,hr) and yield of <u>2</u> (%) |
|---|---|---|
| <u>1a</u>   | <u>2a</u>   | $CH_3CN/20-25/18(65)$ , $CH_2Cl_2/20-25/18(19)$ , neat/20-25/3(35)            |
| <u>1b</u>   | <u>2b</u>   | $CH_3CN/20-25/48(70)$ , $CH_2Cl_2/0/3(73)$ , neat/0/3(81)                     |
| <u>1c</u>   | <u>2c</u>   | $CH_3CN/20-25/12(54)$ , $CH_2Cl_2/20-25/2(60)$ , neat/20-25/3(42)             |
| <u>1d</u>   | <u>2d</u>   | $CH_3CN/20-25/6(94)$ , $CH_2Cl_2/20-25/1(77)$ , neat/20-25/1(78)              |
| <u>1e</u>   | <u>2e</u>   | $CH_3CN/20-25/12(77)$ , $CH_2Cl_2/20-25/12(69)$ , neat/20-25/12(75)           |
| <u>1f</u>   | <u>2f</u>   | $CH_3CN/20-25/15(56)$ , $CH_2Cl_2/20-25/12(46)$ , neat/20-25/0.5(99)          |
| <u>1g</u>   | <u>2g</u>   | $CH_3CN/20-25/4(85)$ , $CH_2Cl_2/20-25/3(73)$ , neat/20-25/12(55)             |
| <u>1h</u>   | <u>2h</u>   | $CH_3CN/20-25/24(50)$ , $CH_2Cl_2/20-25/12(79)$ , neat/20-25/12(78)           |
| <u>1i</u>   | <u>2i</u>   | $CH_3CN/0/2(47)$  |
| <u>1j</u>   | <u>2j</u>   | $CH_3CN/0/2(27)$  |
| <u>1d</u>   |  | $CH_3CN/20-25/24(60)$ , neat/20-25/12(55)                                     |
|  |  | neat/0/0.5(73)  |
| <u>1k</u>   | <u>4</u>  |   |

<sup>a</sup>2-Ethoxy-1,3-dithiane was used instead of 2-ethoxy-1,3-dithiolane.

#### REFERENCES AND NOTES

1. N. K. Kochetkov, E. E. Nifant'ev and V. N. Kulakov, *Doklady Akad. Nauk S. S. S. R.*, **125**, 327 (1959); *Chem. Abstr.*, **53**, 19873 (1959).
2. E. C. Taylor and J. L. LaMattina, *Tetrahedron Lett.*, 2077 (1977).
3. For acid-catalyzed transformation of 2-ethoxy-1,3-dithiolane see S. Tanimoto, T. Miyake and M. Okano, *Bull. Inst. Chem. Res., Kyoto Univ.*, **55**, 276 (1977).
4. All the products in Table 1 (except 2d, 2f and 3) are new compounds. For 2d, 2f and 3, see A. Bargagna, P. Schenone, P. Versace and G. Mallarini, *Pharmazol. Res. Commun.*, 389 (1973); *Chem. Abstr.*, **80**, 128318d (1974), ref 1 and 2 respectively.

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